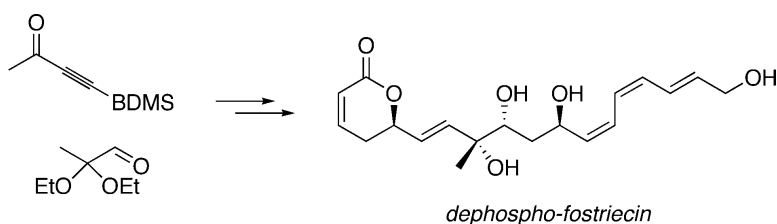


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Dinuclear Asymmetric Zn Aldol Additions: Formal Asymmetric Synthesis of Fostriecin

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Direct asymmetric aldol reactions constitute a powerful methodology for the synthesis of biologically active compounds. With the use of our dinuclear asymmetric zinc complexes we recently demonstrated the feasibility of using alkynyl methyl ketones as donors.¹ One of the advantages of using terminal benzyldimethylsilyl alkynes as donors in this reaction is the potential for employing the resultant adducts directly in subsequent cross-coupling reactions.² The strategic potential offered by these reactions led us to target fostriecin (CI-920, **1**), a cytotoxic phosphate ester isolated from *Streptomyces pulveraceus*.³ The natural product displays potent anticancer activity against leukemia and many other cell lines.⁴ The cytotoxic properties of **1** are attributed to its selective inhibition of protein phosphatase 2A (PP2A).⁵ The absolute and relative stereochemistry of **1** was established by Boger,⁶ who also disclosed the first total synthesis of the natural product⁷ as well as preliminary SAR studies of this molecule.⁸ Several total syntheses have been published,⁹ as well as a number of synthetic approaches.¹⁰ Herein we report our efforts in this area, which culminated in a short and efficient synthesis of dephospho-fostriecin **2**.

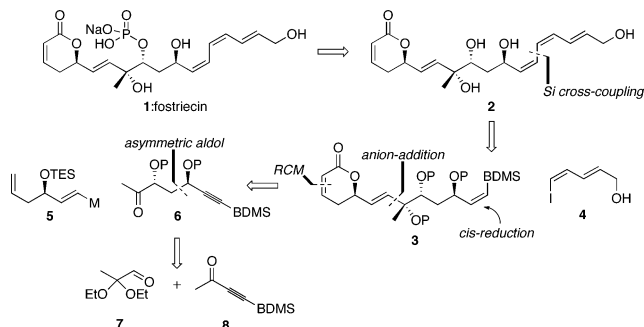
As shown in Scheme 1, we targeted the tetraol **2**, a key intermediate in Boger's synthesis.⁷ We envisioned that **2** could be disconnected between C13 and C14 to give silane **3** and iodide **4**. In the synthetic direction, an alkenylsilane cross-coupling reaction would be employed to forge the labile triene unit.^{2,11} Further disconnection at the C7–C8 bond gives a vinyl metal species **5** and ketone **6**. Finally, ketone **6** would be derived from the aldehyde **7** and ynone **8**, utilizing the direct Zn-catalyzed asymmetric aldol reaction developed in this group.¹

Our synthesis commenced with the preparation of the metalation precursor of **5**. Thus, ethyl propiolate was converted to the β -E-iodoacrolein **9** according to literature procedures (Scheme 2).¹² The labile aldehyde **10** was allylated according to Brown's procedure,¹³ giving the allylic alcohol in high yield and ee (81% and 95%, respectively). Finally, silylation under standard conditions afforded the target vinyl iodide **11**.

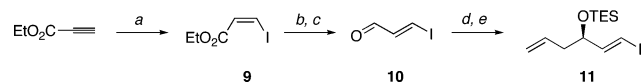
With an efficient route to **11** in hand, attention shifted to the preparation of the C8–C13 portion of fostriecin (**1**) (Scheme 3).

Here we wished to establish the C9 stereochemistry by enantioselective direct aldol reaction as reported recently.¹ Thus, the requisite alkynyl ketone **8** was prepared by the addition of ethynylmagnesium bromide to benzyldimethylchlorosilane (BDMSCl) followed by deprotonation and subsequent acylation. Ketone **12** was treated with aldehyde **7** under our standard Zn-catalyzed direct aldol reaction (3 mol % **13**, 6 mol % Et₂Zn), affording the aldol adduct **14** in an excellent ee (99%). It is worth noting that this reaction could routinely be performed on 50 mmol scale with no deterioration in yield or ee. Use of 5 mol % catalyst bumped the yield to 73%. Next, a diastereoselective reduction was envisaged. Unfortunately, various substrate-controlled methods failed to deliver the desired *anti*-diol in useful yield. Reduction under Noyori's Ru-

Scheme 1. Retrosynthetic Analysis

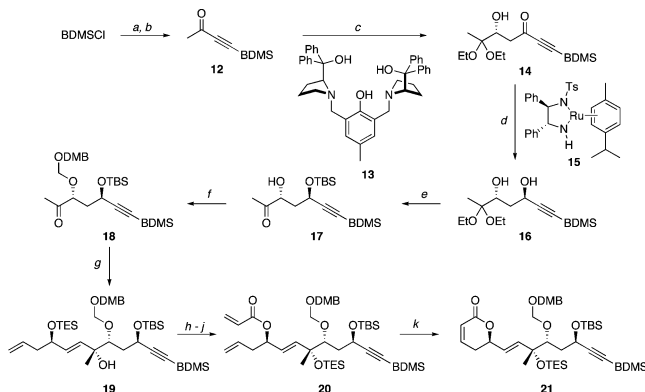


Scheme 2. Synthesis of Vinyl Iodide **11**^a



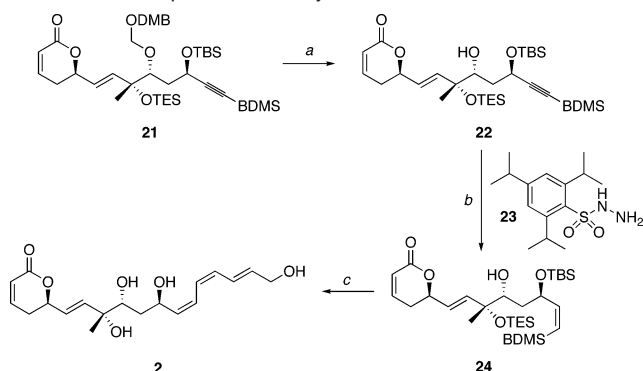
^a Reagents and conditions: (a) NaI, AcOH, 70 °C, (77%); (b) DIBAL-H, CH₂Cl₂, -78 °C; (c) BF₃·Et₂O, CH₂Cl₂, (95:5 *E:Z*); (d) (+)-Ipc₂BOME, allylmagnesium chloride, Et₂O, -90 °C, (49% over 3 steps, 95% ee); (e) TESCl, imidazole, DMF, (95%).

Scheme 3. Synthesis of Alkynyl Silane^a



^a Reagents and conditions: (a) ethynylmagnesium bromide, THF, 0 °C (99%); (b) *n*BuLi, MeCON(Me)OMe, THF, -78 to -15 °C (90%); (c) **13** (3 mol %), Et₂Zn (6 mol %), 4 Å MS, **7**, rt (58–67%, 99% ee); (d) **15** (1 mol %), *i*PrOH, rt (88%, >10:1 dr); (e) TBSCl, Im, DMF, then CSA, Me₂CO (77% 2 steps); (f) DMBCH₂Cl, TBAI (14 mol %), DIEA, DMF, 40 °C (97%); (g) MgBr₂, THF, rt, then **11**, *i*PrMgCl, *s*BuLi, THF, -78 °C (75%, >20:1 dr); (h) HF, MeOH, MeCN, rt (91%); (i) acryloyl chloride, DIEA, CH₂Cl₂ (99%); (j) TESOTf, DIEA, CH₂Cl₂ (99%); (k) Grubbs **1** (10 mol %), CH₂Cl₂, reflux (93%).

catalyzed transfer hydrogenation conditions¹⁴ (cf. **15**) gave the desired alcohol **16** in high yield and selectivity. Selective silylation and hydrolysis of the ketal furnished ketone **17**, which was protected as the DMBO-acetal¹⁵ **18** in high yield with no detectable epimerization at the α -stereocenter. Thus, the stage was set for the introduction of vinyl metal species **5**. A chelation-controlled addition

Scheme 4. Completion of the Synthesis^a

^a Reagents and conditions: (a) DDQ, CH₂Cl₂:H₂O (10:1) (94%); (b) **23**, NaHCO₃, MeOH, rt (53%, 72% brsm); (c) **4**, Pd₂(dba)₃·CHCl₃ (5 mol %), TBAF (4 equiv slow addition), THF, 0 °C to room temperature (54%).

of the corresponding magnesiate species¹⁶ of **11** afforded **19** as a single diastereomer in very good yield (75%). The unsaturated lactone moiety was installed by selective removal of the allylic TES-group, followed by acylation with acrolyl chloride giving **20**, which was subsequently exposed to Grubbs first-generation catalyst, yielding key intermediate **21** (83% over three steps).

With an ample supply of **21** in hand, the stage was set for examination of the final steps toward the target, fostriecin precursor, tetraol **2**. Thus, the acetal protecting group was removed by the addition of DDQ, providing alkyne **22** in 94% yield (Scheme 4).

A *cis* reduction was required to install the vinyl silane. Attempts to effect this transformation were hampered by either poor yield and/or over-reduction. Eventually, using diimide precursor **23**¹⁷ under mildly basic conditions afforded the *cis*-vinyl silane **24**. Finally the assembly of the triene unit called upon a Pd-catalyzed vinyl silane cross-coupling reaction. The susceptibility of the unsaturated lactone toward base complicated this cross-coupling. After much experimentation, it was found that the sensitive triene functionality could be assembled by adding TBAF slowly to a combination of vinyl silane **24**, vinyl iodide **4**,¹⁸ and catalytic amounts of Pd₂(dba)₃·CHCl₃ in THF. As expected, the cross-coupling also led to concomitant deprotection of the silicon protecting groups, giving the dephosphofostriecin **2** in good yield. The conversion of **2** to fostriecin (**1**) has been demonstrated by Boger.⁷

In conclusion, we have completed the synthesis of dephosphofostriecin **2**, and thereby a formal synthesis of fostriecin **1**, in 14 steps for the longest linear sequence and 8.5% overall yield. This work illustrates for the first time the use of the direct asymmetric Zn-catalyzed aldol reaction, and the utility of the corresponding aldol adducts as building blocks for complex molecule synthesis. It also exemplifies the extraordinary utility of the Pd-catalyzed vinyl silane cross-coupling as an alternative to more mainstream Pd-catalyzed cross-coupling reactions, in the synthesis of complex molecules.

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Supporting Information Available: Experimental details and characterization for all new compounds (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>

References

- (1) Trost, B. M.; Fettes, A.; Shireman, B. T. *J. Am. Chem. Soc.* **2004**, *126*, 2660 and references therein to other direct asymmetric aldol reactions.
- (2) Trost, B. M.; Machacek, M. R.; Ball, Z. T. *Org. Lett.* **2003**, *5*, 1895.
- (3) (a) Tunac, J. B.; Graham, B. D.; Dobson, W. E. *J. Antibiot.* **1983**, *36*, 1595. (b) Stampwala, S. S.; Bunge, R. H.; Hurley, T. R.; Willmer, N. E.; Brankiewicz, A. J.; Steinman, C. E.; French, J. C. *J. Antibiot.* **1983**, *36*, 1601. (c) Hokanson, G. C.; French, J. C. *J. Org. Chem.* **1985**, *50*, 462.
- (4) Jackson, R. C.; Fry, D. W.; Boritzki, T. J.; Roberts, B. J.; Hook, K. E.; Leopold, W. R. *Adv. Enzyme Reg.* **1985**, *23*, 193.
- (5) (a) Lewy, D. S.; Gauss, C. M.; Soenen, D. R.; Boger, D. L. *Curr. Med. Chem.* **2002**, *9*, 2005. (b) de Jong, R. S.; Mulder, N. H.; Uges, D. R. A.; Sleijfer, D. T.; Hoppener, F. J. P.; Groen, H. J. M.; Willemse, P. H. B.; van der Graaf, W. T. A.; de Vries, E. G. E. *Br. J. Cancer* **1999**, *79*, 882.
- (6) Boger, D. L.; Hikota, M.; Lewis, B. M. *J. Org. Chem.* **1997**, *62*, 1748.
- (7) Boger, D. L.; Ichikawa, S.; Zhong, W. *J. Am. Chem. Soc.* **2001**, *123*, 4161.
- (8) Buck, S. B.; Hardouin, C.; Ichikawa, S.; Soenen, D. R.; Gauss, C. M.; Hwang, I.; Swingle, M. R.; Bonness, K. M.; Honkanen, R. E.; Boger, D. L. *J. Am. Chem. Soc.* **2003**, *125*, 15694.
- (9) (a) Chavez, D. E.; Jacobsen, E. N. *Angew. Chem., Int. Ed.* **2001**, *40*, 3667. (b) Reddy, Y. K.; Falck, J. R. *Org. Lett.* **2002**, *4*, 969. (c) Miyashita, K.; Ikejiri, M.; Kawasaki, H.; Maemura, S.; Imanishi, T. *Chem. Commun.* **2002**, 742. (d) Miyashita, K.; Ikejiri, M.; Kawasaki, H.; Maemura, S.; Imanishi, T. *J. Am. Chem. Soc.* **2003**, *125*, 8238. (e) Wang, Y. G.; Kobayashi, Y. *Org. Lett.* **2002**, *4*, 4615. (f) Esumi, T.; Okamoto, N.; Hatakeyama, S. *Chem. Commun.* **2002**, 3042. (g) Fujii, K.; Maki, K.; Kanai, M.; Shibasaki, M. *Org. Lett.* **2003**, *5*, 733.
- (10) (a) Just, G.; Oconnor, B. *Tetrahedron Lett.* **1988**, *29*, 753. (b) Liu, S. Y.; Huang, D. F.; Huang, H. H.; Huang, L. *Chin. Chem. Lett.* **2000**, *11*, 957. (c) Cossy, J.; Pradaux, F.; BouzBouz, S. *Org. Lett.* **2001**, *3*, 2233. (d) Kiyotsuka, Y.; Igarashi, J.; Kobayashi, Y. *Tetrahedron Lett.* **2002**, *43*, 2725. (e) Marshall, J. A.; Bourbeau, M. P. *Org. Lett.* **2003**, *5*, 3197. (f) Ramachandran, P. V.; Liu, H. P.; Reddy, M. V. R.; Brown, H. C. *Org. Lett.* **2003**, *5*, 3755.
- (11) For a review, see: Denmark, S. E.; Sweis, R. F. *Acc. Chem. Res.* **2002**, *35*, 835.
- (12) Marek, I.; Meyer, C.; Normant, J. F. *Org. Synth.* **1997**, *74*, 194.
- (13) Brown, H. C.; Jadhav, P. K. *J. Am. Chem. Soc.* **1983**, *105*, 2092.
- (14) (a) Matsumura, K.; Hashiguchi, S.; Ikariya, T.; Noyori, R. *J. Am. Chem. Soc.* **1997**, *119*, 8738. (b) Haack, K.-J.; Hashiguchi, S.; Fujii, A.; Ikariya, T.; Noyori, R. *Angew. Chem., Int. Ed. Engl.* **1997**, *36*, 285.
- (15) Kigoshi, H.; Suenaga, K.; Mutou, T.; Ishigaki, T.; Atsumi, T.; Ishiwata, H.; Sakakura, A.; Ogawa, T.; Ojika, M.; Yamada, K. *J. Org. Chem.* **1996**, *61*, 5326.
- (16) (a) Inoue, A.; Kitagawa, K.; Shinokubo, H.; Oshima, K. *J. Org. Chem.* **2001**, *66*, 4333. (b) Kitagawa, K.; Inoue, A.; Shinokubo, H.; Oshima, K. *Angew. Chem., Int. Ed.* **2000**, *39*, 2481. (c) Knochel, P.; Dohle, W.; Gommernann, N.; Kneisel, F. F.; Kopp, F.; Korn, T.; Sapountzis, I.; Vu, V. A. *Angew. Chem., Int. Ed.* **2003**, *42*, 4302.
- (17) Cusack, N. J.; Reese, C. B.; Risius, A. C.; Roozpekar, B. *Tetrahedron* **1976**, *32*, 2157.
- (18) Vinyl iodide **4** was prepared from ester **9**, via DIBAL-H reduction, Horner–Wadsworth–Emmons reaction and subsequent DIBAL-H reduction, which afforded **4** in 33% overall yield, see Supporting Information for more details.

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